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Acta Cryst. (1993). C49, 1031-1032

# Structure of 2-(2'-Thienyl)pyridine at 193 K

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(Received 15 September 1992; accepted 10 November 1992)

## Abstract

The molecular geometry of the thiophene and pyridine rings is very similar to other derivatives. Both rings are planar and almost in the same plane. The slight twist between the two planes arises from the intramolecular short contact between S and N. The C4'-C5' bond between the thiophene and pyridine rings is a partial double bond as a result of conjugation.

## Comment

2-(2'-Thienyl)pyridine (Figs. 1 and 2) (Aldrich) crystallizes from methanol solution in the non-centrosymmetric space group  $P2_12_12_1$ . The rotation of 4.48° around the C2-C2' bond joining the thiophene and pyridine rings results in a conformationally chiral molecule. The twist between the two rings may be attributed to the short intramolecular contact between S and N (N···S = 2.933) Å; the sum of the radii = 3.35 Å). The bond C2–C2' [1.469(3) Å] between the thiophene and pyridine rings is shorter than a C-C single bond (1.54 Å), implying a resonance between the rings. The C5'-C4' and C3'-C2' distances of 1.360(5) and 1.369(4) Å, respectively, are slightly longer than a double bond, and the C4'-C3' distance of 1.407 (4) Å is shorter than a single bond. The C-C and C-S bonds in the thiophene ring are in good agreement with those in 6-phenyl-4-(2-thienyl)-1H-pyrimidin-2-one (Tenon, Ebby, Degny, Viani & Bodot, 1988), 2-(5-methyl-3-thienyl)-2H,5Hpyrazolo[4,3-c]quinolin-3-one (Nakai, 1990) and 2-(5methyl-2-thienyl)-2,5-dihydro-3H-pyrazolo[4,3-c]quinolin-3-one (Shiro, 1990). The molecular geometry of the six-membered ring is very similar to that found in 4aminopyridine (Chao & Schempp, 1977), 2-amino-4-

0108-2701/93/051031-02\$06.00





Fig. 1. View of C<sub>9</sub>H<sub>7</sub>NS showing the labelling of the non-H atoms. Thermal ellipsoids are shown at 30% probability levels; H atoms are drawn as small circles of arbitrary radii.



Fig. 2. Packing diagram viewed down the b axis.

Experimental	
Crystal data	

Z = 4

C<sub>9</sub>H<sub>7</sub>NS Mo  $K\alpha$  radiation  $M_r = 161.22$  $\lambda = 0.71073 \text{ Å}$ Orthorhombic Cell parameters from 30 reflections  $P2_{1}2_{1}2_{1}$ a = 5.769 (3) Å  $\theta = 18.66 - 30.06^{\circ}$  $\mu = 0.32 \text{ mm}^{-1}$ b = 8.793 (3) Å T = 193 Kc = 15.695 (6) Å Rectangular block V = 796.2 (6) Å<sup>3</sup>  $0.64\,\times\,0.29\,\times\,0.24$  mm  $D_x = 1.345 \text{ Mg m}^{-3}$ White

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Data collection

Duiu concention		List
Nicolet P3 diffractometer	$h = 0 \longrightarrow 8; 0 \longrightarrow 8$	dina brar
$\omega$ scans	$k = -12 \rightarrow 0; 0 \rightarrow 12$	557
2237 measured reflections	$l = -21 \rightarrow 0; 0 \rightarrow 21$	Inte
1099 independent reflections	4 standard reflections	2111
1023 observed reflections	monitored every 96	2110
$[F>4.0\sigma(F)]$	reflections	
$R_{\rm int} = 0.0086$	intensity variation: 1.28%	Det
$\theta_{\rm max} = 55.0^{\circ}$		Cha
Refinement		Dest
Refinement on F	$\Delta \rho_{\rm max} = 0.38 \ {\rm e} \ {\rm \AA}^{-3}$	2i Goli
Final $R = 0.0349$	$\Delta \rho_{\rm min} = -0.16 \ {\rm e} \ {\rm \AA}^{-3}$	2
wR = 0.0499	Extinction correction:	Kvi
S = 1.9613	$F^* = F[1 + 0.002\chi F^2]$	Nak
1023 reflections	$(\sin(2\theta))^{-1/4}$	She
129 parameters	Extinction coefficient:	ic
All H-atom parameters re-	0.0012	Shir
fined	Atomic scattering factors	Spe
Calculated weights $w =$	from International Tables	
$1/(-2/E) + 0.000/E^2$	for X-ray Crystallogra-	Ien
$1/[\sigma(r) + 0.0004r]$	nhy (1074 Vol. IV. Table	, C
$(\Delta/\sigma)_{\rm max} = 0.065$	2.3.1)	

Data collection: SHELXTL-Plus (Sheldrick, 1991). Cell refinement: SHELXTL-Plus. Data reduction: SHELXTL-Plus. Program(s) used to solve structure: SHELXTL-Plus. Program(s) used to refine structure: SHELXTL-Plus. Molecular graphics: SHELXTL-Plus. Software used to prepare material for publication: PLATON (Spek, 1990).

# Table 1. Fractional atomic coordinates and equivalent isotropic thermal parameters $(Å^2)$

 $U_{\text{eq}} = \frac{1}{3} \sum_{i} \sum_{j} U_{ij} a_{i}^{*} a_{i}^{*} \mathbf{a}_{i} . \mathbf{a}_{j}.$ 

		-	•		
	x	у		z	$U_{eq}$
S	0.3566 (1)	0.04553	8 (8)	0.04855 (4)	0.0411 (2)
N	0.3627 (4)	-0.1937	(3)	0.1787 (1)	0.0355 (6)
C2'	0.1411 (4)	-0.0868	(3)	0.0655 (1)	0.0279 (6)
C3'	-0.0322 (5)	-0.0699	(3)	0.0066 (2)	0.0361 (7)
C4'	0.0111 (5)	0.0481	(3) –	0.0518 (2)	0.0420 (8)
C5'	0.2145 (6)	0.1218	(3) -	0.0372 (2)	0.0423 (8)
C2	0.1616 (4)	-0.1991	(3)	0.1344 (1)	0.0286 (6)
C3	-0.0133 (5)	-0.3025	(4)	0.1520 (2)	0.0391 (8)
C4	0.0202 (6)	-0.4077	(4)	0.2165 (2)	0.0465 (9)
C5	0.2279 (5)	-0.4045	(4)	0.2618 (2)	0.0434 (8)
C6	0.3867 (5)	-0.2933	(4)	0.2388 (2)	0.0397 (8)
	Table 2. C	Geometric	parame	eters (Å, °)	
S-C5'		1.712 (3)	C2—C3		1.386 (4)
S-C2'		1.723 (3)	C2—N		1.353 (3)
C5'C4'		1.360 (5)	NC6		1.294 (4)
C4'-C3'		1.407 (4)	C3C4		1.385 (4)
C3'-C2'		1.369 (4)	C4—C5		1.393 (4)
C2′C2		1.469 (3)	C5—C6		1.387 (4)
C5' - S - C5'	22'	92.3 (1)	C2'C2-	C3	121.9 (2)
SC5'C	C4'	111.0 (2)	N-C2-	C3	123.0 (2)
C5'-C4'-	-C3'	113.3 (3)	C2C3	-C4	118.8 (3)
C4'-C3'-	-C2'	112.9 (2)	C3C4-	-C5	118.6 (3)
C3'C2'-	C2	128.9 (2)	C4-C5-	-C6	116.7 (3)
C3'C2'-	S	110.5 (2)	C5C6-	-N	126.5 (3)
S-C2'-C	22	120.6 (2)	C6—N	C2	116.3 (2)
C2'C2-	-N	115.1 (2)			

SHS acknowledges the support of the Robert A. Welch Foundation (grant F-0017).

0108-2701/93/051032-04\$06.00

of structure factors, anisotropic thermal parameters, H-atom coores and complete geometry have been deposited with the British Li-Document Supply Centre as Supplementary Publication No. SUP 7 (8 pp.). Copies may be obtained through The Technical Editor, national Union of Crystallography, 5 Abbey Square, Chester CH1 , England. [CIF reference: ST1039]

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Acta Cryst. (1993). C49, 1032-1035

# Structures of Tetradecyltrimethylammonium Salicylate Monohydrate (1) and Hexadecyltrimethylammonium Salicylate Monohydrate (2)

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(Received 10 August 1992; accepted 9 December 1992)

#### Abstract

(1)  $[(C_{14}H_{29})(CH_3)_3N]^+ .C_7H_5O_3^- .H_2O$ : the tetradecyl chain is fully extended in a zigzag form; the cation, anion and the water molecule are packed in layers with the tetradecyl chains parallel within a layer and antiparallel in alternate layers; the water molecule is hydrogen bonded to one of the carboxylate O atoms of the anion. (2)  $[(C_{16}H_{33})(CH_3)_3N]^+.C_7H_5O_3^-.H_2O$ : the crystal packing, bond lengths and bond angles are all very similar to those of (1).

## Comment

It has been reported recently that aqueous solutions of (1) and (2) show interesting viscoelastic and spinnable properties (Imae, Hashimoto & Ikeda, 1990;

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